

Amino acids, peptides and proteins

amino acids

basic building units

peptides

significant biological effects

proteins

basic constituent of all living cells

→ **foods**



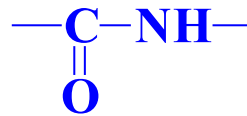
basic nutrient

amino acids

functional groups: NH₂ COOH

compounds of amino acids bounded by

peptide bond



N-substituted amino acid amides

peptides

2-100 amino acids

proteins

> 100 amino acids (typically hundreds to thousands)

Amino acids

> 700 amino acids in plant and animal tissues

as **free amino acids** in food

1%

as **building units of peptides, proteins and other compounds**

99%

structure

always at least 1 group



free, substituted

1 group



additional functional groups

1. hydroxyl



2. sulfhydryl (thiol)



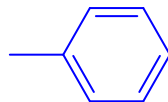
3. sulfide



4. guanidyl



5. phenyl



classification

- according to distance of amino group from carboxylic group

		n
2-amino acids	α -amino acids	0
3-amino acids	β -amino acids	1
4-amino acids	γ -amino acids	2
5-amino acids	δ -amino acids	3
6-amino acids	ϵ -amino acids	4

- according to their occurrence

➤ found in all organisms (invariable)

20 basic (coded, proteinogenic) **amino acids**

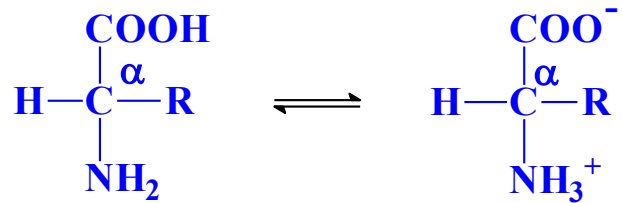
➤ found in only some organisms (variable)

2 proteinogenic amino acids - **selenocystein, pyrrolysin**

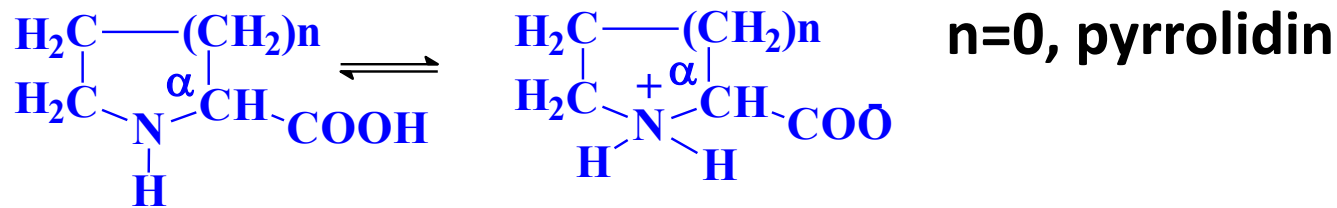
about 700 other amino acids(free, bounded)

basic amino acids

19 α -amino acids with primary amino group $-\text{NH}_2$



1 α -amino acid with secondary amino group $-\text{NH}-$ (proline)



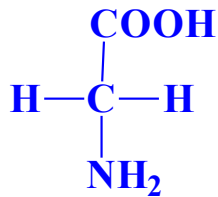
19 amino acids = chiral compounds, L-isomers

Trivial names	Systematic names	Three letter symbols	One letter symbols
glycine	aminoacetic	Gly	G
L-alanine	L-2-aminopropionic	Ala	A
L-valine	L-2-amino-3-methylbutyric	Val	V
L-leucine	L-2-amino-4-methylvaleric	Leu	L
L-isoleucine	L-2-amino-3-methylvaleric	Ile	I
L-serine	L-2-amino-3-hydroxypropionic	Ser	S
L-threonine	L-2-amino-3-hydroxybutyric	Thr	T
L-cysteine	L-2-amino-3-merkaptopropionic	Cys	C
L-methionine	L-2-amino-4-methylthiobutyric	Met	M
L-aspartic acid ^{a)}	L-aminosuccinic	Asp	D
L-glutamic acid ^{b)}	L-2-aminoglutaric	Glu	E
L-asparagine ^{a)}	L-2-amino-4-carbamoylbutyric	Asn	N
L-glutamine ^{b)}	L-2-amino-5-carbamoylvaleric	Gln	Q
L-lysine	L-2,6-diaminohexanoic	Lys	K
L-arginine	L-2-amino-5-guanidylvaleric	Arg	R
L-histidine	L-2-amino-3-(4-imidazolyl)propionic	His	H
L-phenylalanine	L-2-amino-3-phenylpropionic	Phe	F
L-tyrosine	L-2-amino-3-(4-hydroxyphenyl)propionic	Tyr	Y
L-tryptophan	L-2-amino-3-(3-indolyl)propionic	Trp	W
L-proline	L-pyrrolidin-2-carboxylic	Pro	P
	^{a)} aspartic acid and asparagine	Asx	B
	^{b)} glutamic acid and glutamine	Glx	Z

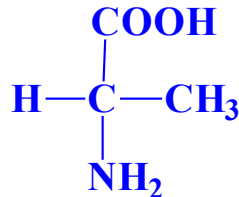
classification of basic amino acids

according the structure of side-chain and functional groups

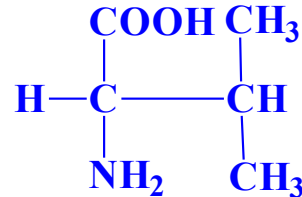
aliphatic amino acids with **nonsubstituted** chain



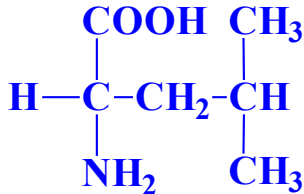
glycine



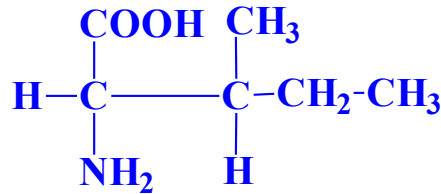
alanine



valine

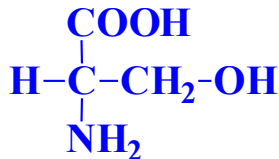


leucine

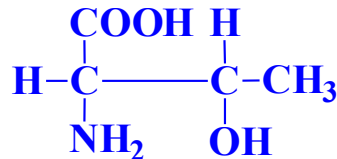


isoleucine

alifatic **hydroxy**amino acids

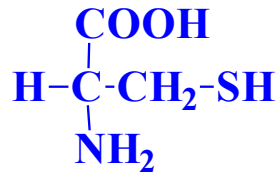


serine

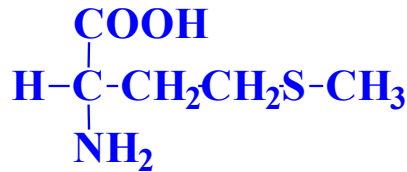


threonine

aliphatic **sulphur** amino acids

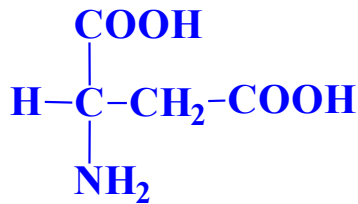


cysteine

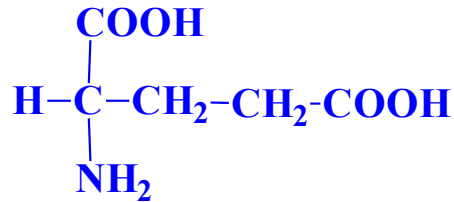


methionine

with **carboxyl group** in the side-chain (monoaminodicarboxylic acids, acidic)

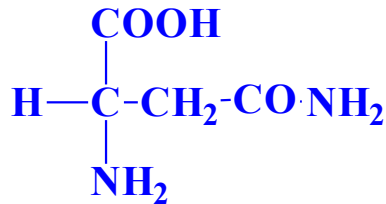


aspartic acid

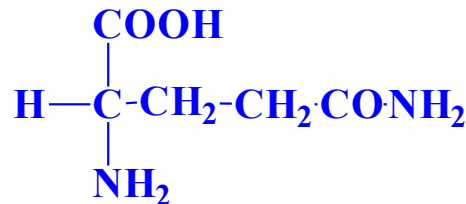


glutamic acid

their **monoamides** (with carboxamide group in the side-chain)



asparagine



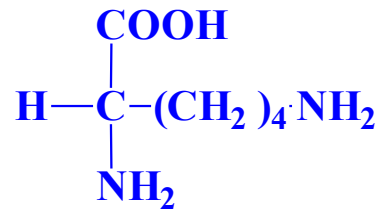
glutamine

with **basic** groups in the side-chain

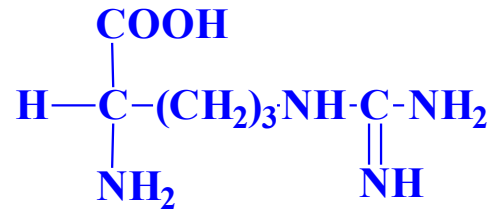
second amino group (lysine)

guanidylic group (arginine)

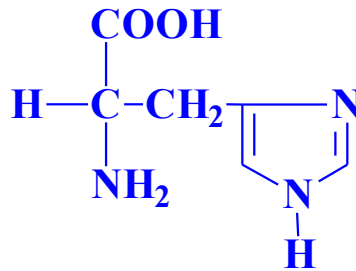
imidazole cycle (histidine)



lysine

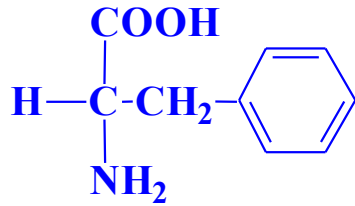


arginine



histidine

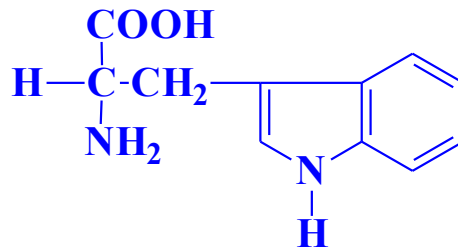
with aromatic (heterocyclic) side-chain



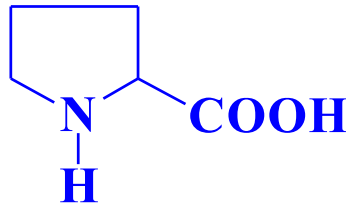
phenylalanine



tyrosine



tryptophan (derivative of indole)



proline (derivative of pyrrolidine)

classification in biochemistry

according to polarity of the side-chain and its ionic forms (in neutral medium) – related to protein interactions


- nonpolar, hydrophobic: Val, Leu, Ile, Phe, Tyr, Met, Pro;
to some extent Gly, Ala, Trp (amphiphilic)
- polar, hydrophilic: Ser, Thr, Cys, Asp, Glu, Asn, Gln, Lys, Arg, His



by ionic forms of the side chain in organisms :

- neutral (without electric charge): most amino acids
- acidic (negative electric charge): Asp, Glu
- basic (positive electric charge): Lys, Arg, His

classification according to importance in human nutrition

- **essential:** Val, Leu, Ile, Thr, Met, Lys, Phe, Trp  only from food
- **semiessential:** Arg, His by fast growing organisms (children)
- **nonessential:** others

coded amino acids in human body synthesized:

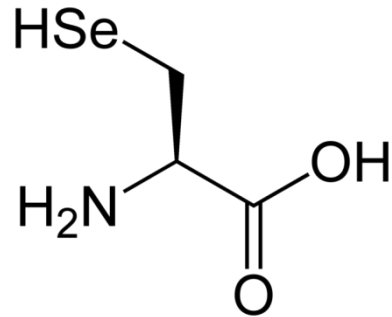
- from other amino acids
- from glucose
- from fatty acids etc.

limiting = amino acid, which is in the diet / commodity present in relatively less amount (based on recommended daily intake)



determines the nutritional value of food / commodity
only essential amino acids !

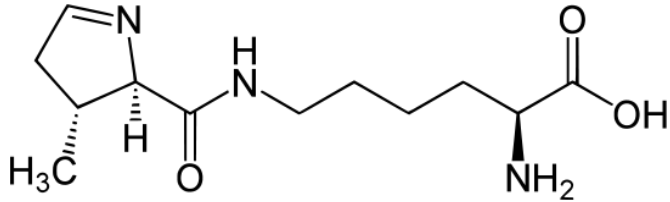
21. proteinogenic amino acid - selenocysteine



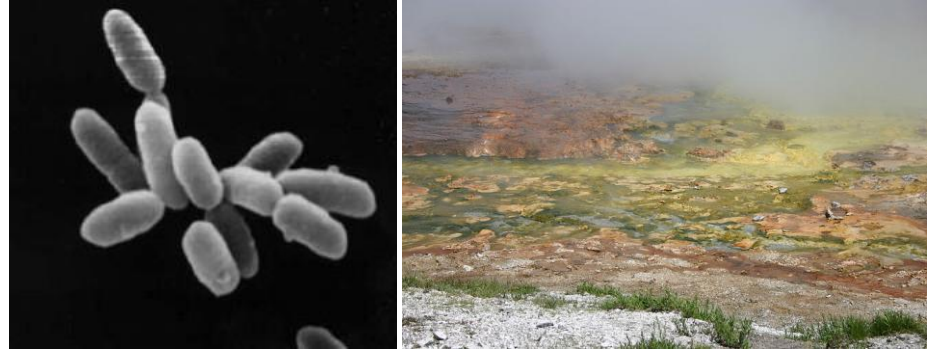
Selenocysteine (Se-Cys)

- higher reduction potential than Cys - in proteins with antioxidant activity
- in enzymes (glutathione peroxidase, formate dehydrogenases, glycine reductase, some hydrogenase)

22. proteinogenic amino acid - pyrrolysine



pyrrolysine



Layers of thermophilic archaea in Yellowstone National Park

Archaea - an ancient form of life - is estimated to be 3.5 billion years on Earth
- known primarily as an extremophile organisms growing also at high temperature, extreme pH or high salt content

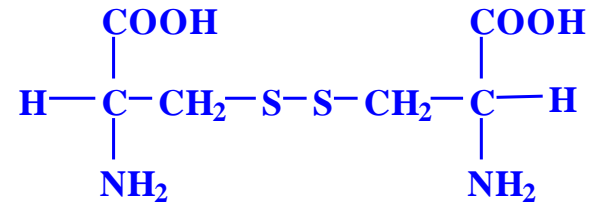
In the food industry enzymes amylase, galactosidase and pululanase isolated from archaea *Pyrococcus* genus are used to produce food (e.g. milk and whey with low amount of lactose) - even at temperatures above the boiling point still retains their catalytic function

Methanobrevibacter smithii - most common archeon in the human intestine
- represents about one tenth of all prokaryotic organisms in the gut

derivatives of proteinogenic amino acids

- 20 basic amino acids = ~ 90% amino acids in foods
- their derivatives - formed by posttranslational modifications

L-cystine (CySSCy)



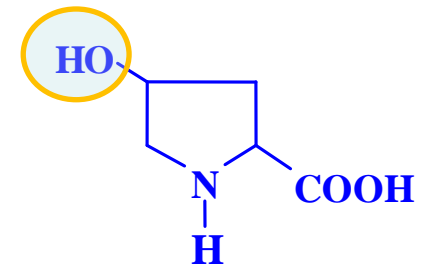
L-4-hydroxyproline

(L-4-hydroxypyrrolidin-2-carboxylic acid, Hyp)

important structural component of connective tissue, **collagen** and **gelatin** protein (content about 12%)

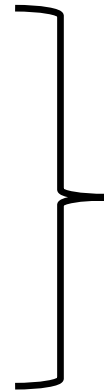
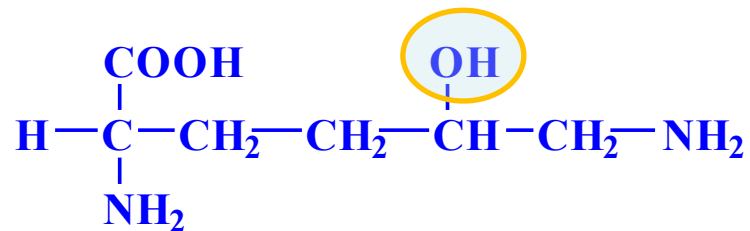


amount in meat products correlates with the amount of lower-quality materials such as skin or cartilage



L-3-hydroxyproline

L-5-hydroxylysine



to a small extent accompany
L-4-hydroxyproline in
collagen

3-methylhistidine

minority in myofibrillar meat proteins (actin and myosin)

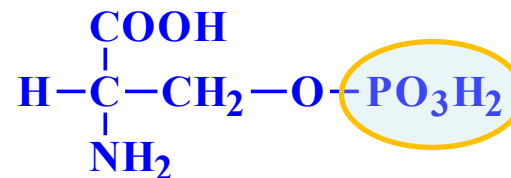
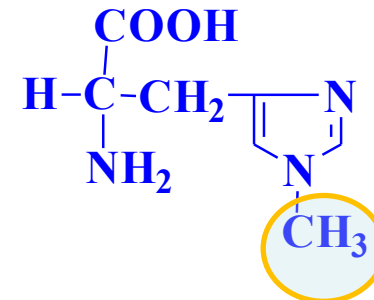
critereon for determining the quality of ingredients in meat products

O-fosfoserin

fosfoproteins (phosvitin in eggs)

glycoproteins (κ -casein)

phospholipids (cereal germs)



Amino acid	Content in proteins (%)		
	Beef muscle*	Beef collagen**	Wheat proteins***
Gly	5.0	<u>31.10</u>	2.8
Ala	4.0	11.0	2.0
Ser	5.4	3.8	4.0
Thr	5.3	2.0	2.3
Pro	6.0	<u>11.8</u>	<u>11.5</u>
Hypro	<u>0.0</u>	<u>10.1</u>	<u>0.0</u>
Val	5.8	2.1	2.6
Ileu	6.3	1.2	3.1
Leu	8.0	2.8	6.2
Phe	4.5	1.6	2.9
Tyr	3.1	0.3	1.1
Trp	1.2	<u>0.0</u>	1.0
Cys (+CySSCy)	1.1	<u>0.0</u>	2.3
Met	<u>3.2</u>	0.5	<u>1.6</u>
Asp(+ AspNH ₂)	6.0	5.0	3.7
Glu(+GluNH ₂)	<u>15.4</u>	7.6	<u>34.6</u>
Arg	7.2	4.9	2.9
His	1.9	0.6	2.0
Lys	<u>7.6</u>	2.6	1.9
Hyllys	<u>0.0</u>	<u>0.6</u>	<u>0.0</u>

*actin, myosin

**collagen

***glutenin, gliadin

The content of amino acids in cereals and pseudocereals (in g based on 16 g of nitrogen)

AA	wheat	rye	barley	oat	rice	corn	millet	buckwheat	amaranth
Ala	3.6	4.3	4.0	4.5	6.0	7.5	7.9	4.7	3.4
Arg	4.6	4.6	4.7	6.3	8.3	4.2	5.3	9.8	7.4
Asx	4.9	7.2	5.7	7.7	10.3	6.3	8.0	8.9	8.3
Cys	2.5	1.9	2.3	2.7	1.1	1.6	2.4	2.4	1.4
Glx	29.9	24.2	23.6	20.9	20.6	18.9	18.6	17.3	15.4
Gly	3.9	4.3	3.9	4.7	5.0	3.7	3.8	5.0	8.7
His	2.3	2.2	2.1	2.1	2.5	2.7	2.4	2.1	2.3
Ile	3.3	3.5	3.6	3.8	3.8	3.7	4.1	3.4	3.6
Leu	6.7	6.2	6.7	7.3	8.2	12.5	9.6	5.9	5.3
Lys	2.9	3.4	3.5	3.7	3.8	2.7	3.4	3.8	5.0
Met	1.5	1.5	1.7	1.7	2.3	1.9	2.5	1.5	1.8
Phe	4.5	4.4	5.1	5.0	5.2	4.9	4.8	3.8	3.6
Pro	9.9	9.4	10.9	5.2	4.7	8.9	6.1	4.3	3.6
Ser	4.6	4.3	4.0	4.7	5.4	5.0	4.9	5.0	7.1
Thr	2.9	3.3	3.3	3.3	3.9	3.6	3.9	3.6	3.5
Trp	0.9	1.9	0.9	1.1	0.8	0.7	2.0	1.4	1.5
Tyr	3.0	1.9	3.1	3.3	3.5	3.8	3.2	2.4	3.4
Val	4.4	4.8	5.0	5.1	5.5	4.8	5.5	6.7	4.3
Sum EAA ¹⁾	32.8	31.6	35.8	37.1	38.5	40.2	41.1	34.8	28.4
Sum AA ²⁾	96.5	92.0	94.6	93.3	101.2	97.5	98.1	93.3	89.4
EAAI (%) ³⁾	68	75	78	79	76	55	67	76	76
AAS (%) ⁴⁾	44	46	54	57	57	41	53	51	54
Limiting AA	Lys	Trp Ile	Lys Leu	Ile Lys	Ile Lys	Lys	Lys	Lys, Ile	Lys, Ile

¹⁾ EAA = essential amino acids, ²⁾ AA = amino acids, ³⁾ EAAI = index of essential AA, ⁴⁾ AAS = score of AA

The content of amino acids in legumes, oil seeds and nuts (in g based on 16 g of nitrogen)

AA	Soy	Lens	Pea	Beans	Sunflower	Peanuts	Sesame	Walnut	Hazelnut
Ala	4.3	4.3	4.1	4.2	4.2	3.9	4.5	4.1	4.2
Arg	7.2	8.7	9.5	5.7	8.0	11.2	12.1	12.3	15.0
Asx	11.7	11.6	11.0	12.0	9.3	11.4	8.2	8.3	7.2
Cys	1.3	0.9	1.1	0.8	1.5	1.2	1.8	0.5	0.4
Glx	18.7	16.6	16.1	14.8	21.8	18.3	19.4	20.1	20.5
Gly	4.2	4.2	4.0	3.8	5.4	5.6	4.9	7.0	8.7
His	2.5	2.7	2.3	2.8	2.3	2.4	2.4	2.0	1.8
Ile	4.5	4.3	4.3	4.2	4.2	3.4	3.6	3.9	6.2
Leu	7.8	7.6	6.8	7.6	6.4	6.4	6.7	7.5	6.2
Lys	6.4	7.2	7.5	7.2	3.6	3.5	2.7	1.6	2.9
Met	1.3	0.8	0.9	1.1	1.9	1.2	2.8	1.3	0.8
Phe	4.9	5.2	4.6	5.2	4.4	5.0	4.4	4.1	3.6
Pro	5.5	4.3	3.9	3.6	4.5	4.4	3.7	4.7	5.6
Ser	5.1	5.3	4.3	5.6	4.3	4.8	4.7	6.1	9.6
Thr	3.9	4.0	4.1	4.0	3.7	2.6	3.6	2.7	2.7
Trp	1.3	1.5	1.4	1.4	1.4	1.0	1.0	1.0	1.1
Tyr	3.1	3.3	2.7	2.5	1.9	3.9	3.1	3.1	3.7
Val	4.8	5.0	4.7	4.6	5.2	4.2	4.6	4.4	6.4
Sum EAA ¹⁾	39.3	39.8	38.2	38.6	34.1	32.4	34.8	26.5	33.1
Sum AA ²⁾	98.5	97.4	93.4	90.9	93.9	94.2	94.7	94.5	106.7
EAAI (%) ³⁾	62	41	50	47	93	69	63	60	35
AAS (%) ⁴⁾	47	31	37	34	56	43	43	24	22
Limiting AA	Met Val	Met Trp	Met Trp	Met Trp	Met Lys	Met Ile	Lys Ile	Met Lys	Met Lys

¹⁾ EAA = essential amino acids, ²⁾ AA = amino acids, ³⁾ EAAI = index of essential AA, ⁴⁾ AAS = score of AA

other amino acids (nonproteinogenic)

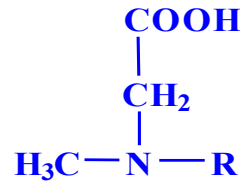
secondary metabolites

- products of various metabolic processes
- biosynthetic precursors of some nitrogenous compounds
- with specific functions (neural mediators and hormones)
- toxic compounds (plant protection from predators)
- storage and transport form of nitrogen

products of microorganisms

products of chemical transformation of proteins (isomerization, hydrolysis) or free amino acids (isomerization)

N-substituted α -amino acids

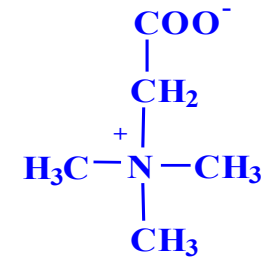


N-methylglycine (sarkosine, R = H)

by degradation of creatine

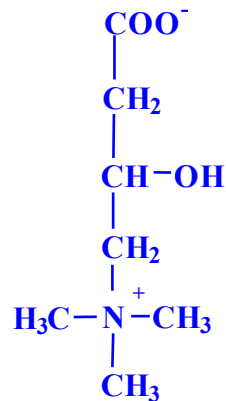
N,N-dimethylglycine

part of pangamic acid
(vit B₁₃, yeast, cereals)



N,N,N-trimethylglycine (betaine)

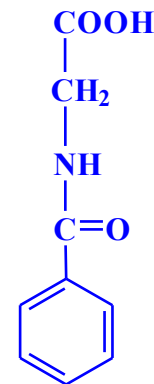
in sugar beet



L-carnitine (3-hydroxy-4-trimethylaminobutyrate, vitamin B_t)

foods of animal origin, food supplements

metabolismus lipidů

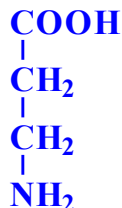


hippurová kyselina (N-benzoylglycin)

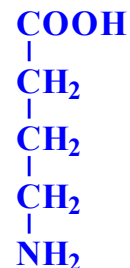
from Gly by detoxication of benzoic acid

(plants, milk)

β -aminoacids and γ -aminoacids



β -alanine (3-aminopropionic acid)
decarboxylation of Asp (by lyases)
part of pantothenic acid (vit. B₅)



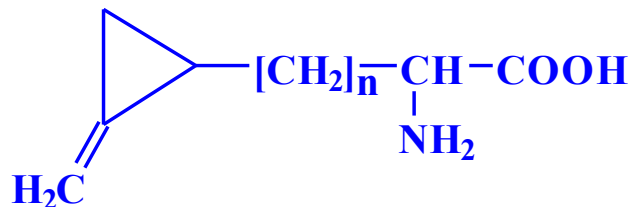
γ -aminobutyric (4-aminobutyric) acid (GABA)
decarboxylation of Glu
brain tissue, inhibitor of transmission of nerve impulses

alicyclic amino acids



1-aminocyclopropan-1-carboxylic acid

fruits (apples, pears,...)
ethylene precursor (stimulates maturation)

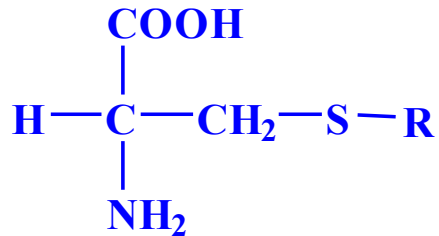


2-amino-3-(2-methylcyclopropyl)propionic acid
(hypoglycine)

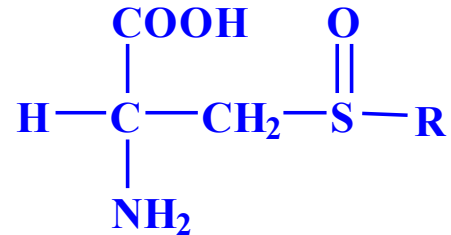
in unripe ackee fruit (Africa, Jamaica,...)
hypoglycaemic and other toxic effects



sulphurous amino acids



S-alk(en)yl L-cysteines



S-alk(en)yl-L-cysteine sulfoxides

S-methylcysteine sulfoxide (methiin, R = CH₃)

cruciferous (cabbage), Lily (garlic, etc.), toxic for ruminants

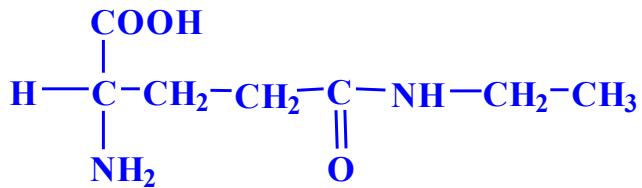
S-allyl-cysteine sulfoxide (alliin, R = CH₂-CH=CH₂)

garlic aroma

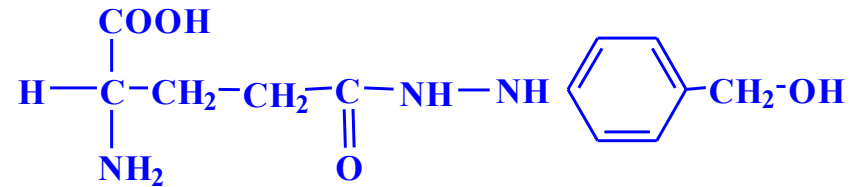
S-(prop-1-en-1-yl)cysteine sulfoxide (isoalliin, R = CH=CH-CH₃)

onion

acidic amino acids



***N*-ethyl-L-glutamine (L-theanin, L-ethanin)**
tea leaves



agaritine (0.03 - 0.173% f.w.)

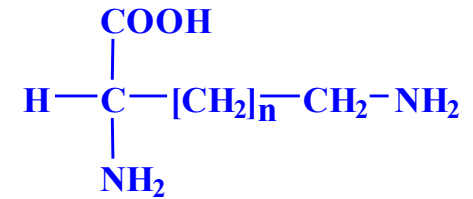
antiviral, hydrazine-derivative mycotoxin
carcinogen that occurs in mushroom
species of the genus *Agaricus*
decompose readily upon cooking (up to
90% reduction)



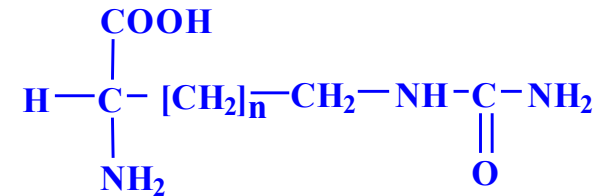
Agaricus bisporus

basic amino acids and related compounds

L-ornithine (L-2,5-diaminovaleric acid, n = 2)
arginine biosynthesis, ornithine (urea) cycle



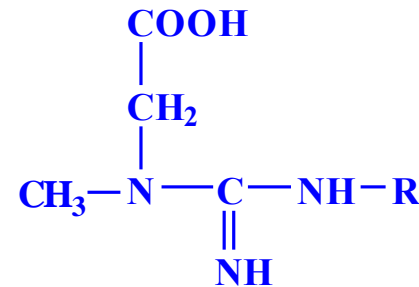
L-citrulline (n = 2, karbamoylderivát ornithinu)



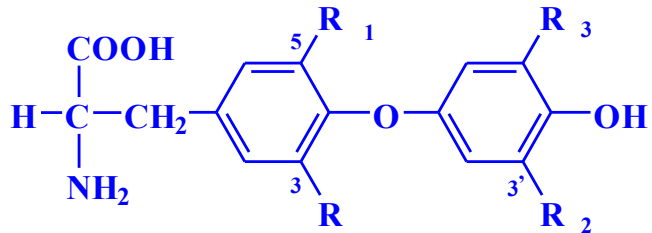
intermediate of formation of urea, nitrogen reserve in plants, the transport form of nitrogen

creatine-phosphate

muscle tissue (3-6 g/kg), an energy reserve for muscle activity and regeneration of ATP
creatine (R = H)



aromatic amino acids

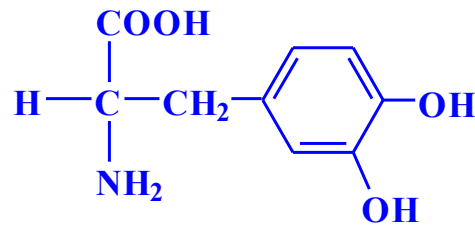


thyronine, $R = R^1 = R^2 = R^3 = H$

trijodthyronine, $R = R^1 = R^2 = I, R^3 = H$

tetraiodthyronine (thyroxine), $R = R^1 = R^2 = R^3 = I$

hormonal effects



L-3,4-dihydroxyphenylalanine (DOPA)

precursor of brown to black pigment **melanin**

enzymatic browning reaction

PHYSIOLOGY AND NUTRITION

varied diet  **essential AA supplying is sufficient**

possibility of enriching foods with essential (limiting) AA

- **Lys** (low content in cereals and vegetable proteins in general)
- **Met** (lower content in meat and milk proteins)
- **Thr** (low content in wheat and rye proteins)
- **Trp** (lower content of milk caseins, proteins, corn and rice)

human nutrition and livestock feed (performance)

0,05-0,2% as additives in feedingstuffs (mainly Lys, Met) – for farmed animals

PHYSICAL PROPERTIES

Acidobasic properties

in physiological values of pH

α -carboxylic groups, α -amino groups and α -imino groups dissociated

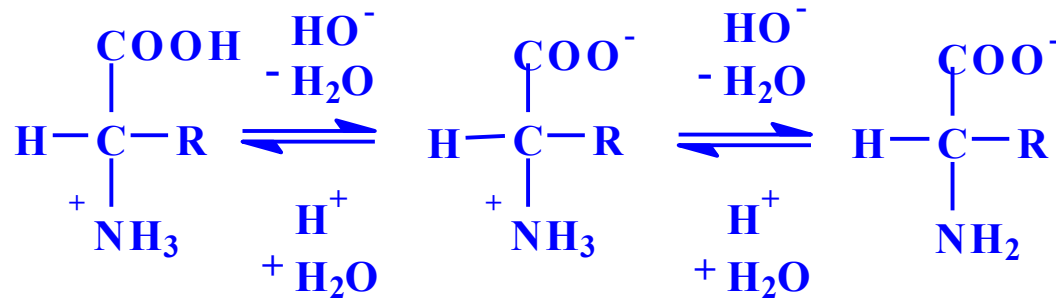
- amfolytes (amphoteric electrolytes)
- form a so-called inner salt (dipolar, amphoteric ion, amfion, zwitterion)

AA currently carries a positive and negative electrical charge
resulting electric charge of the molecule is zero

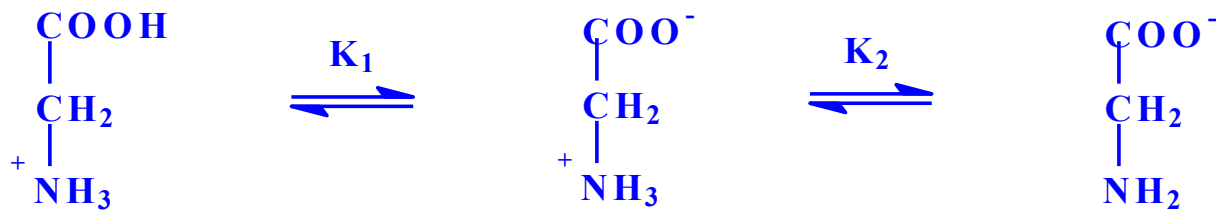
amfion - predominant form of almost all amino acids in the
physiological environment of animal and plant tissues

depending on pH of medium - except **amfions** also **cations** (if acting as base) and **anions** (if acting as acid)

disociation of AA depending on the pH of the medium



disociation of glycine depending on the pH of the medium:



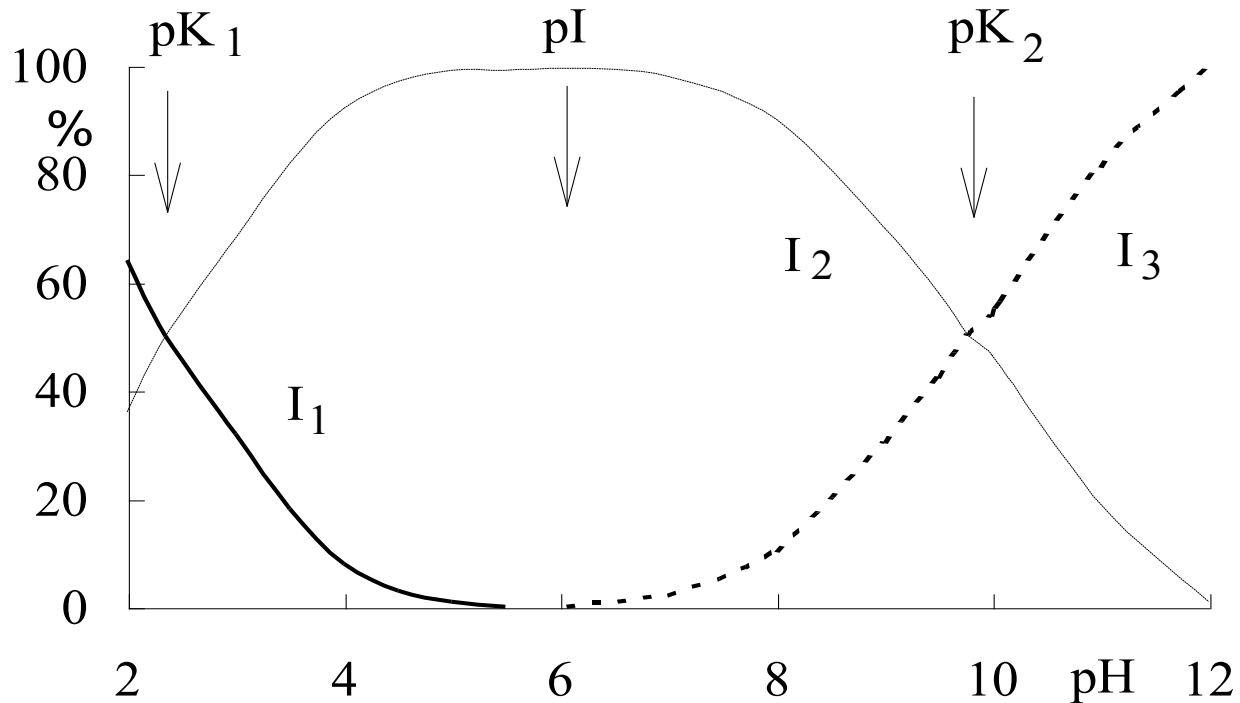
ion I₁ (cation)
 free charge +1
 pH < 2

ion I₂ (amfion)
 free charge 0
 pH ≈ 6

ion I₃ (anion)
 free charge -1
 pH > 10

- maximum concentration of amfions in pH = pI (isoelectric point):

$$pI = \frac{pK_1 + pK_2}{2} = 6,1$$

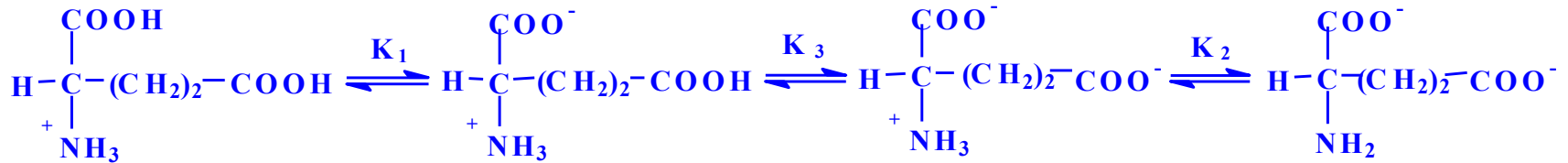


dependence of ionic forms of glycine at pH

———— = cation (I₁) — — — — = amfion (I₂) = anion (I₃)

acidic amino acids

disociation of Glu depending on pH:



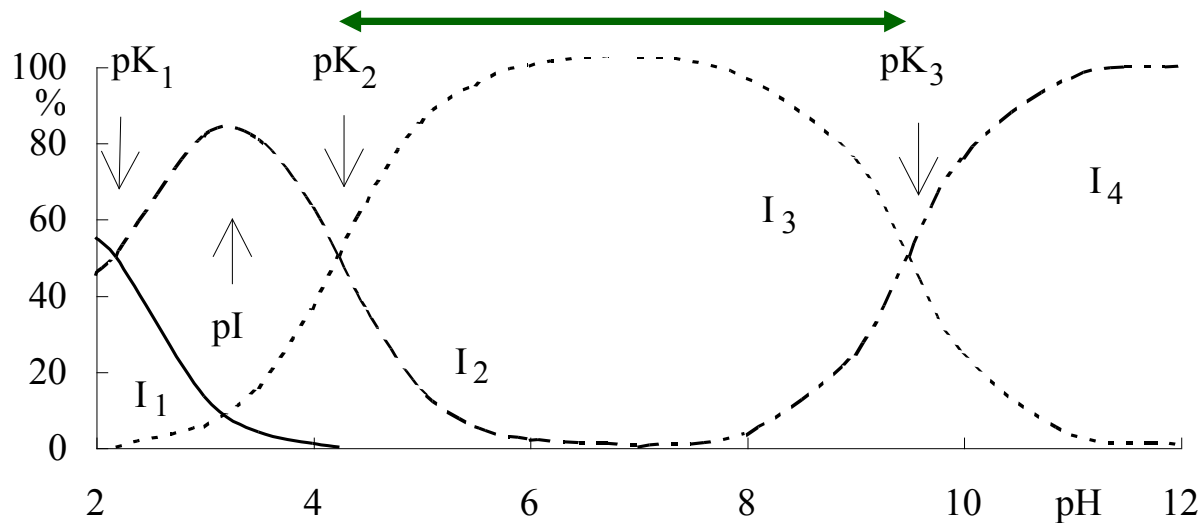
ion I_1 (cation)
free charge +1
pH < 2

ion I_2 (amfion)
free charge 0
pH \approx 3

ion I_3 (anion)
free charge -1
pH \approx 7

ion I_4 (anion)
free charge -2
pH > 10

desirable organoleptic properties



dependence of ionic forms of glutamic acid on pH

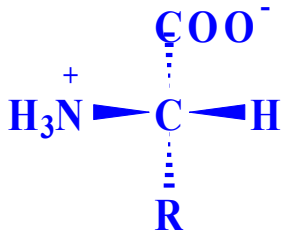
— = cation (I_1) - - - - = amfion (I_2) = anion (I_3) - · - · - · = anion (I_4)

optical activity

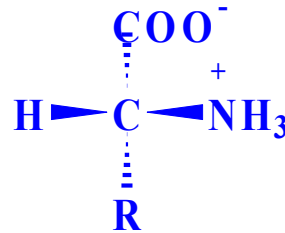
glycine = exception

other amino acids = chiral carbon atom C_{α}

2 optical isomers (enantiomers)



L-amino acid
(*S*)-amino acid



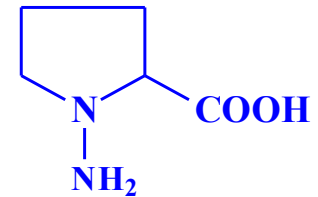
D-amino acid
(*R*)-amino acid

- **amino acids with L-configuration = (*S*)-stereoisomers**
derived from L-glyceraldehyd
exception: L-cysteine = (*R*)-stereoisomer
(priority of R=CH₂SH is higher than -COOH)
- **amino acids with D-configuration = (*R*)-stereoisomers**
derived from D-glyceraldehyd

amino acids with D-configuration



- sporadic in nature in biologically active peptides of plants and animals
- in peptidoglycans of cell walls of microorganisms
- free compounds (eg. linatin in flax seeds)

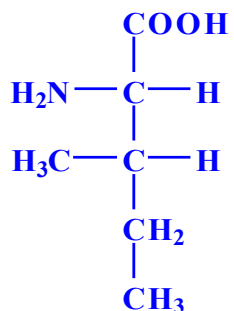


in fresh, unprocessed foods mostly of microbial origin

other origin

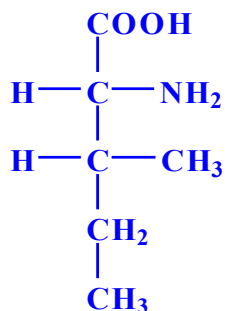
- non-enzymatic isomerization of L-amino acids in food processed at high temperatures in an alkaline pH (abiogenic D-amino acids)

diastereoisomers



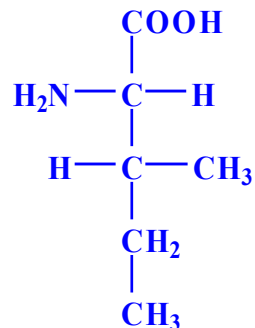
L-isoleucine

(2S, 3S)-isoleucine



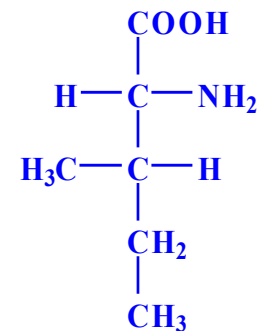
D-isoleucine

(2R, 3R)-isoleucine



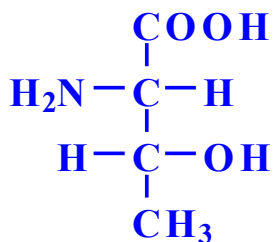
L-*allo*-isoleucine

(2S, 3R)-isoleucine



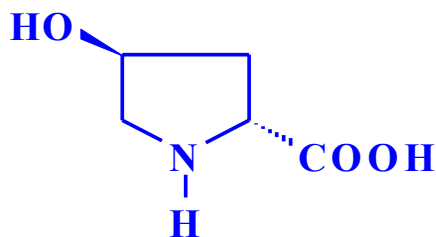
D-*allo*-isoleucine

(2R, 3S)-isoleucine



L-threonine

(2S, 3R)-threonine



L-4-hydroxyproline

(2S, 4R)-4-hydroxyproline

ORGANOLEPTIC PROPERTIES

sensory active substances, particularly taste

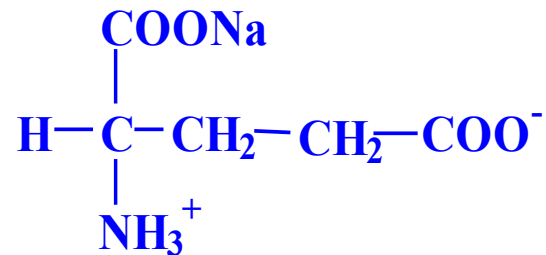
- **sweet** (Gly, Ala, Thr, Pro)
- **sour** (Asp, Glu)
- **bitter** (with hydrophobic side chains: Leu, Ile, Phe, Tyr, Trp)
- **indifferent** (all others)

umami (in Japanese = delicious) full, meaty, bringing a sense of satisfaction in the mouth

unique organoleptic properties = Glu,

resp. **sodium-hydrogen glutamate**

intensifier meaty taste (soups, sauces)



additive (E 621), in U.S.A. abbreviated as MSG (Mono Sodium Glutamate)

CHEMICAL PROPERTIES

reactions: (isomerization)

elimination

addition

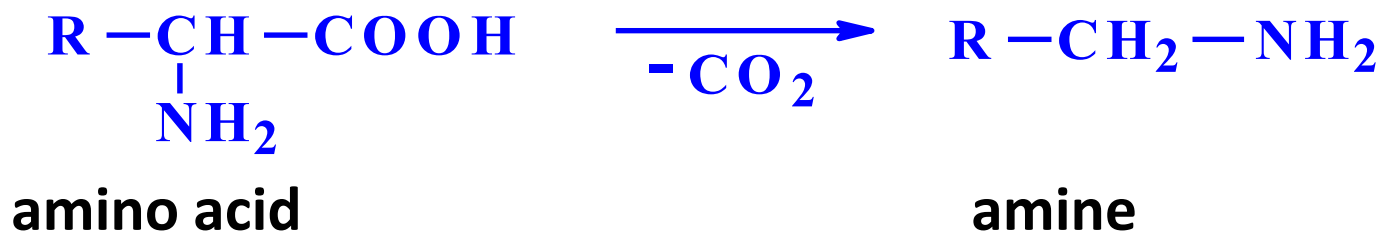
oxidation

formation of flavour compounds

(amines, aldehydes, alcohols, sulphurous compounds)

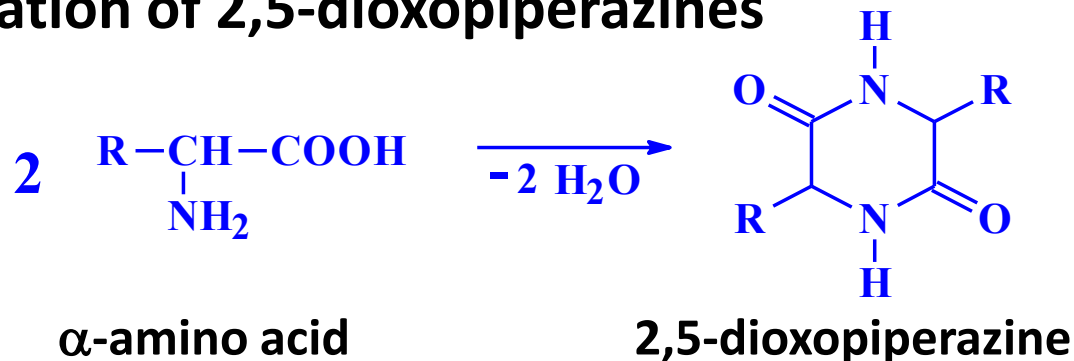
Elimination reactions

decarboxylation (elimination of CO₂)



elimination of ammonia and water

formation of 2,5-dioxopiperazines



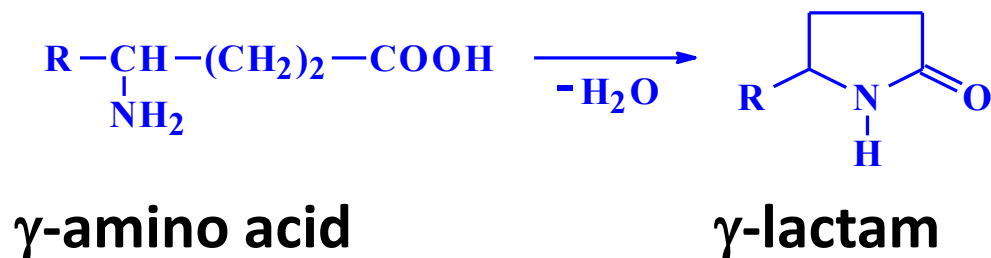
Gly etc.

formation of alk-2-enic acids



Asp

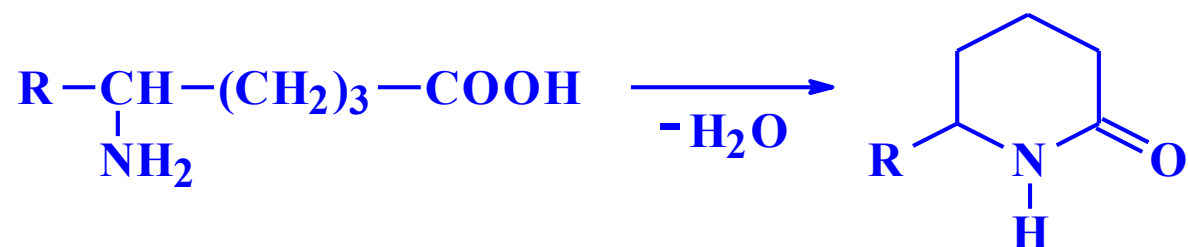
formation of γ -lactams



Glu

A lactam (of the words lactone + amide) is a cyclic amide.

formation of δ -lactams

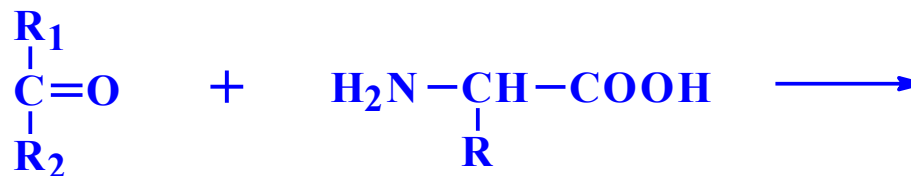


δ -amino acid

δ -lactam

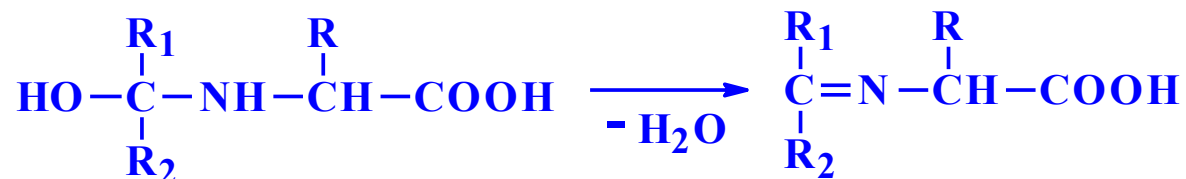
Addition reactions

reaction with aldehydes and ketones



carbonyl compound

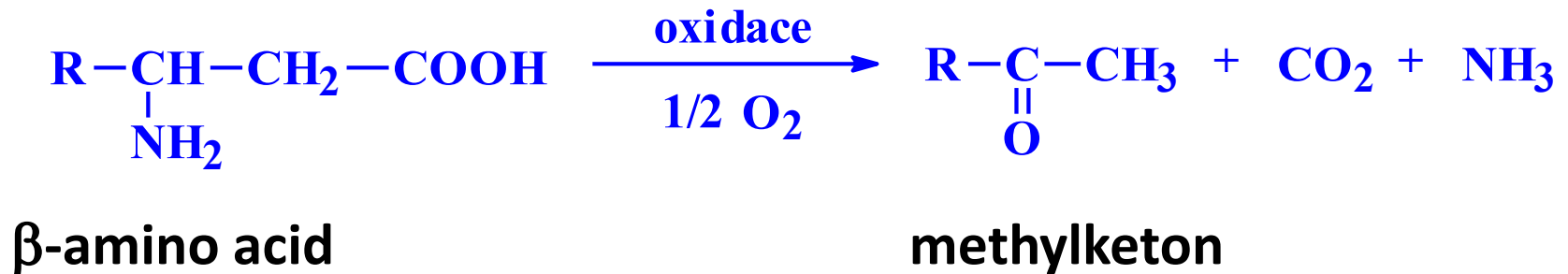
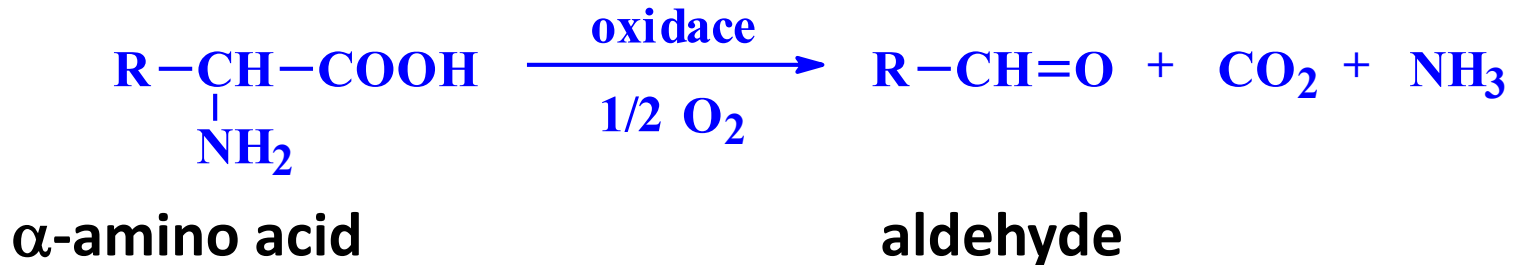
α -amino acid



imin (Schiff's base)

OXIDATION REACTIONS

Strecker's degradation (oxidative decarboxylation)



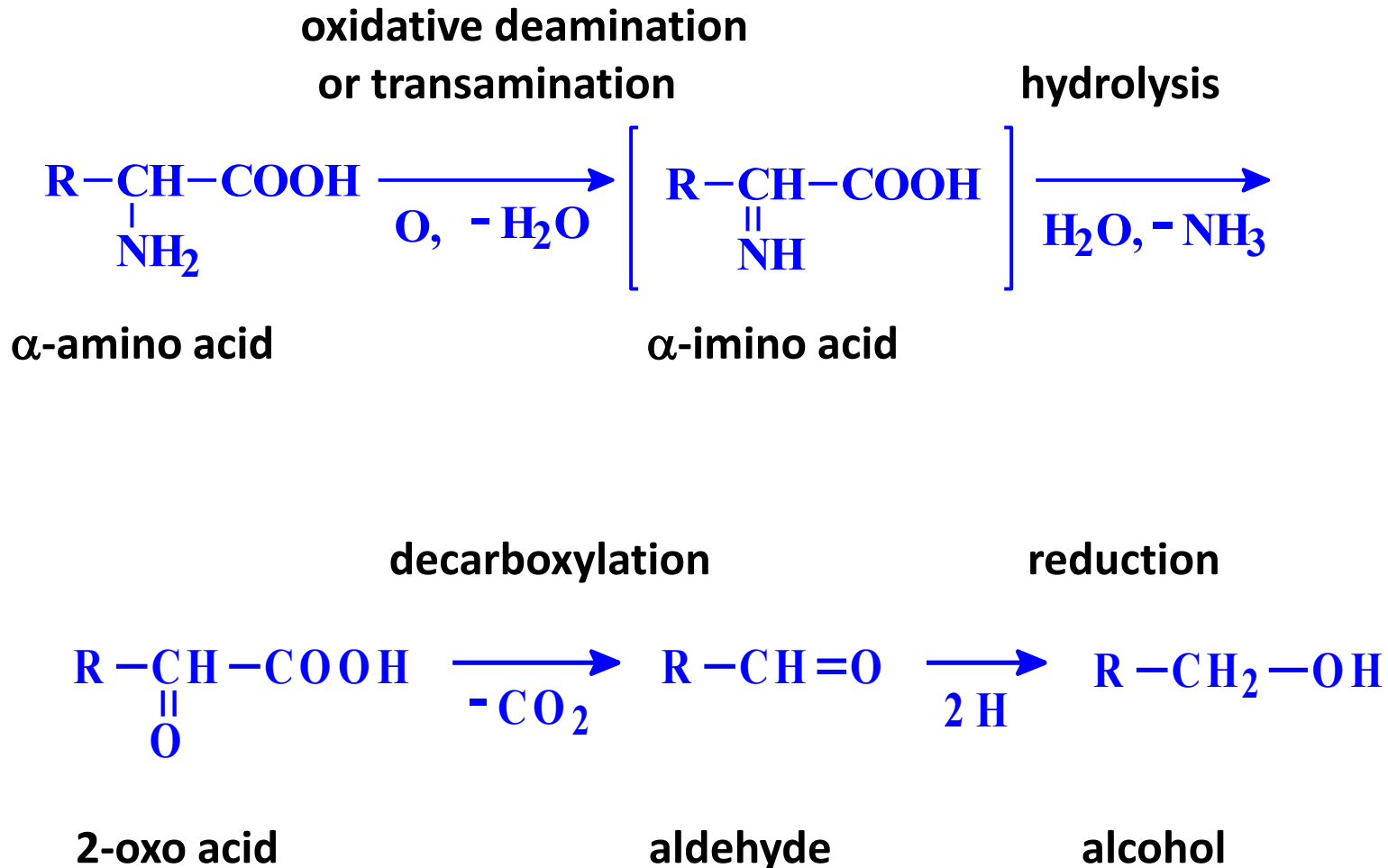
Strecker's degradation aldehydes

Amino acid	Product of degradation
Gly	methanal (formaldehyd)
Ala	ethanal (acetaldehyd)
Thr	propanal
α - a γ -aminobutyric acid	propanal
Val	2-methylpropanal
Leu	3-methylbutanal
Ile	2-methylbutanal
Met	methional
Phe	2-phenylethanal (phenylacetaldehyd)
norvalin	Butanal
norleucin	Pentanal
Cys	2-merkptoacetaldehyd
Orn	4-aminobutanal
Asp	oxalacetic acid (acetaldehyd)
Glu	2-oxoglutaric acid (propanal)

oxidative deamination and transamination

(enzymes, the mechanism like Strecker degradation)

2-ketoacids → aldehydes → alcohols



the word Fusel is German for "bad liquor"

- aldehydes mainly in fruits and vegetables
- alcohols in alcoholic beverages(fusel alcohols)

influence on flavour

alcohol	amino acid
propan-1-ol	Thr
butan-1-ol	Thr
2-methylbutan-1-ol (isobutanol)	Val
(S)-2-methylbutan-1-ol	Ile
3-methylbutan-1-ol	Leu
2-phenylethanol	Phe
tyrosol	Tyr
tryptofol	Trp

influence on flavour

Some beverages, such as rum, whisky (especially Bourbon) and traditional ales and ciders, are **expected** to have relatively high concentrations of non-hazardous alcohols **as part of their flavor profile**

However, in other beverages, such as vodka and lagers, the presence of other alcohols than ethanol is considered **a fault**.